

LP Tuesday, October 8th and Wednesday, October 9th, 2019

Unit 3.2

Group	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
Period	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1	H																	He
2	Li	Be											B	C	N	O	F	Ne
3	Na	Mg											Al	Si	P	S	Cl	Ar
4	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
6	Cs	Ba	La*	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
7	Fr	Ra	Ac**	Rf	Db	Sg	Bh	Hs	Mt	Uun	Uuu	Uub	Uuq			Uuh		Uuo

*Lanthanides	58	59	60	61	62	63	64	65	66	67	68	69	70	71
	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
**Actinides	90	91	92	93	94	95	96	97	98	99	100	101	102	103
	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

CALIFORNIA CONTENT STANDARDS:

Standardized Test Prep, Chapter 5, Electrons in Atoms

- 6.1.1 Explain how elements are organized in a periodic table
- 6.1.2 Compare early and modern periodic tables
- 6.1.3 Identify three broad classes of elements

Stamp: **Atoms/Molecules**

Copy New Vocabulary List Chapter 6

Copy Element List: # 1

Quiz on Elements # 1 List Thursday, Friday

Lecture: **Chapter 6.1**
PPT: Element Classes

Class Work: **Periodic Table: Part 1**
Periodic Table Trans 16

Home Work: **None**